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web this experiment was being performed so that dibenzalacetone could be synthesized from benzaldehyde and acetone this experiment was performed to show how a ketone and an aldehyde could be added together through the aldol condensation the aldol condensation is extremely important because it can form a  $\beta$  hydroxy aldehyde or

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web dibenzalacetone was synthesized in the lab from benzaldehyde and acetone under basic conditions the gc ms was used to relate the molar mass of the product thin layer chromatography was used to visualize the relative rf factor for each isomer uv vis instrument showed three peaks that were assigned to each isomer

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web an aldol reaction was performed by combining acetone and benzaldehyde to make dibenzalacetone the percent yield of dibenzalacetone was found to be 73 which is a good percent yield the melting point of dibenzalacetone was around the range of 104-110 °C this helped conclude that dibenzalacetone had a trans-trans-*E,E* isomer

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