



SAMAGRA SHIKSHA, KERALA
TERMINAL EVALUATION 2022-'23
CHEMISTRY

E 906-Ch

Standard: IX

Time : 1½ Hours
Total Score : 40

Instructions

- The first 15 minutes is cool off time. You may use the time to read and plan your answers.
- Answer the questions only after reading the instructions and questions thoroughly.
- Score and time are to be considered while answering.

Answer any FOUR questions from 1 to 5. Each question carries 1 score. (4 x 1 = 4)

1. Which shell among the following has the highest energy? (1)
(K, L, M, N)
2. How are the atoms called after they get charge? (1)
3. Name of the scientist who suggested the planetary model of atom is (1)
4. Find out the relationship and fill up.
Laws of electrolysis : Michael Faraday
Law of Conservation of mass : (1)
5. The isotope of hydrogen used in nuclear reactors is (1)

Answer any FOUR questions from 6 to 10. Each question carries 2 scores. (4 x 2 = 8)

6. Some statements regarding atom are given below. Find the correct statements (2)
 - a) In an atom number of protons and electrons are not equal.
 - b) Atom is the smallest particle that can take part in a chemical reaction.
 - c) Atoms of different elements have same atomic number.
 - d) In an atom the entire mass is concentrated at its nucleus.
7. The element 'X' forms a diatomic molecule with double bond.
 - a) How many electrons are there in the outer most shell of 'X'? (1)
 - b) How many pairs of electrons are shared for this bond formation? (1)

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