



PEUGEOT

Engine management

Model: 106 1.1/1.6 • 107 1.6 • 206/CC 1.1/1.4 16V/1.6/1.6 16V/2.0 • 206 1.6
207 1.4 16V/1.6/2.0 • 406 2.0/2.2/3.0 • 406 Coupe 2.0/2.2/3.0
607 1.6/2.0/2.2/3.0 • 607 2.0/2.2/3.0 • 806 1.6 • 807 2.0/2.2/3.0
Partner 1.1/1.4/1.6 • Boxer II 1.6 • Expert 1.0

EOBD type	Fault location	Probable cause
P1327	Ignition coil, camshaft position (CMP) sensor - control circuit	Wiring, ignition coil, CMP sensor, ECM
P1329	Knock sensor (KS) 1 - signal	Wiring, knock sensor (KS)
P1330	Knock sensor (KS) 2 - signal	Wiring, knock sensor (KS)
P1331	Knock sensor (KS) 3 - signal	Wiring, knock sensor (KS)
P1332	Knock sensor (KS) 4 - signal	Wiring, knock sensor (KS)
P1336	Ignition system - misfire detected	Wiring, ignition/fuel system
P1337	Ignition system, cylinder 1 - misfire detected	Wiring, ignition/fuel system
P1338	Ignition system, cylinder 2 - misfire detected	Wiring, ignition/fuel system
P1339	Ignition system, cylinder 3 - misfire detected	Wiring, ignition/fuel system
P1340	Ignition system, cylinder 4 - misfire detected	Wiring, ignition/fuel system
P1358	Ignition coil A - malfunction	Wiring, ignition coil, ECM
P1359	Ignition coil B — malfunction	Wiring, ignition coil, ECM
P1454	Exhaust gas recirculation (EGR) system - range/ performance problem	Wiring, hose leak/blockage, EGR valve/solenoid
P1455	Exhaust gas recirculation (EGR) system - range/ performance problem	Wiring, hose leak/blockage, EGR valve/solenoid
P1461	Engine control module (ECM) - adaptation to EGR valve	Adaptation to EGR valve incomplete/not carried out, EGR valve tight/sticking
P1462	Engine control module (ECM) - adaptation to EGR valve	Adaptation to EGR valve incomplete/not carried out, EGR valve tight/sticking
P1507	Engine coolant blower motor, rationality check - malfunction	Wiring, engine coolant blower motor, ECM
P1517	Immobilizer - programming	Incorrect programming, transponder, multifunction control module, combination switch
P1518	Brake pedal position (BPP) switch - signal	Wiring, BPP switch
P1519	Engine coolant blower motor - supply circuit/earth circuit malfunction	Wiring, engine coolant blower motor, ECM
P1520	Fuel gauge tank sensor - signal	Wiring, fuel gauge tank sensor
P1523	Idle speed control (ISC) actuator - circuit malfunction	Wiring, no/incorrect signal
P1524	Idle speed control (ISC) actuator - circuit malfunction	Wiring, no/incorrect signal
P1525	Idle speed control (ISC) actuator - circuit malfunction	Wiring, no/incorrect signal
P1526	Alternator - circuit	Wiring, alternator, battery discharged, multifunction control module
P1527	Alternator - circuit	Wiring, alternator, battery discharged, multifunction control module
P1528	Alternator - circuit	Wiring, alternator, battery discharged, multifunction control module
P1529	Alternator - circuit	Wiring, alternator, battery discharged, multifunction control module
P1543	Air conditioning (AC) - request signal	Wiring, multifunction control module, AC refrigerant pressure sensor
P1555	Exhaust gas recirculation (EGR) system - range/ performance problem	Wiring, hose leak/blockage, EGR valve/solenoid
P1603	Engine coolant temperature (ECT) sensor - range/ performance problem	Wiring, ECT sensor, multifunction control module

Peugeot 406 Fault Code Manual

Xiang Xie

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing
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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2 which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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