

Paper I – Multiple Choice Questions (10 marks)

1. Energy is stored in a battery and in a box of matches. Which type of energy is stored in each of them?

	a battery	a box of matches
A.	chemical	chemical
B.	chemical	internal (thermal)
C.	electrical	chemical
D.	electrical	internal (thermal)

2. A gasoline-powered vehicle goes up a hill from point X to point Y. What energy transformation must occur in the car's engine?

- A. Chemical energy into mechanical energy
 B. Electrical energy into chemical energy
 C. Heat energy into chemical energy
 D. Mechanical energy into nuclear energy

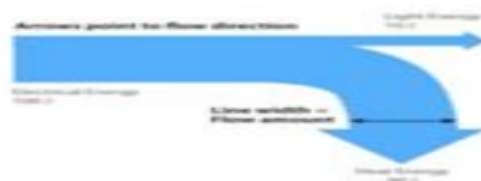
3. Which of these resources will most likely be depleted first?

- A. Wind C. Solar energy
 B. Fossil fuels D. Water

4. What type of wasted energy is given off a light bulb?

- A. Elastic C. Light
 B. Heat D. Kinetic

5. What do we call the graph which represents energy?



- A. Scatter Graph C. Sankey Diagram
 B. Stinky Diagram D. Bar Chart

6. Where does most of the energy go for this transformation?



- A. heat C. electrical
 B. light D. joules

7. This diagram shows the energy transferred by an electric heater in one second:



How much energy is wasted each second?

- A. 12J C. 50J
 B. 38J D. 88J

8. What is an advantage to solar energy?

- A. it pollutes
 B. it is gathered 24 hours a day
 C. it works well no matter where you are in the world
 D. it is renewable

9. What law states that energy cannot be created or destroyed but can only be changed from one form to another?

- A. law of conservation of energy
 B. law of conservation of matter
 C. law of gravity
 D. law of proportions

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CliffsAP Chemistry, 4th Edition Bobrow Test Preparation Services, 2011-09-26 Your complete guide to a higher score on the AP Chemistry exam Why CliffsAP Guides Go with the name you know and trust Get the information you need fast Written by test prep specialists Contents include Introduction overview of the test and how it is scored proven strategies for each type of question Review of topics tested atom periodic table bonding geometry hybridization stoichiometry gases liquids and solids thermodynamics solutions equilibrium acids and bases kinetics redox nuclear chemistry organic chemistry and writing reactions The Labs feature 20 multiple choice questions multiple free response questions on each topic with answers on each topic with answers and explanations scoring rubrics and 2 full length practice exams Structured like the actual exam Complete with answers and explanations AP is a registered trademark of the College Board which was not involved in the production of and does not endorse this product *CliffsNotes AP Chemistry* Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book

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[Problems and Problem Solving in Chemistry Education](#) Georgios Tsaparlis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner

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