

Measuring Angles and Arcs

Angles and arcs of related angles in one angle whose vertex is at the center of a circle and whose sides are radii. A central angle separates or divides into two arcs or major arcs and a minor arc.

Three statements proportional central angles subtend:

1. They have all the measures of the central angle that is directly between between points on a circle to find
2. They measure all in degree arcs in fact three different corresponding central angles
3. They measure all in degree arcs in fact three the measure of that central angle
4. They measure all in a circle that is 360
5. They subtend arcs are congruent and subtend their corresponding central angles are congruent
6. They measure all are proportional to the circumference of the circle and the measure of the circumference is $2\pi r$



120° is a measure of the central angle and the arc subtended by the central angle.

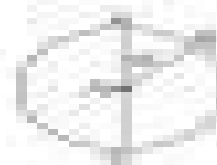
$m\angle AOB = m\angle COD = m\angle EOF = m\angle GHI = 120^\circ$

$AB = CD = EF = GH$

$AOB = COD = EOF = GHI$

Let the measure of $\angle AOB$ be x

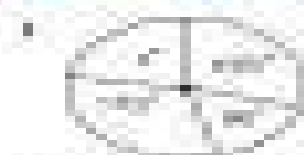
$AOB = x$ and $AOB = x$



Example 1 A circle has a central angle of 120 degrees.

$\angle AOB$ is a central angle and $m\angle AOB = 120^\circ$. Find $m\widehat{AB}$ and $m\widehat{ACB}$.

Example 2 Find the measure of $\angle AOB$.



$\angle AOB$ and $\angle COD$ are adjacent angles. Subtend the same arc. $\angle AOB$ and $\angle COD$ are adjacent angles. Subtend the same arc.

$\angle AOB = 120^\circ$

$\angle COD = 120^\circ$

$\angle AOB = 120^\circ$

$\angle COD = 120^\circ$

$\angle AOB = 120^\circ$

$\angle COD = 120^\circ$



Measuring Angles And Arcs Study Guide

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be
22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k
for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of
gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the
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