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web the half-life of a specific radioactive isotope is constant; it is unaffected by conditions and is independent of the initial amount of that isotope. Consider the following example: suppose we have 100.0 g of tritium, a radioactive isotope of hydrogen.

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web half-life chemistry questions with solutions. Q1: An isotope of caesium Cs 137 has a half-life of 30 years. If 1.0 g of Cs 137 disintegrates over a period of 90 years, how many grams of Cs 137 would remain? A: a) 1.25 g, b) 0.125 g, c) 0.00125 g, d) 12.5 g. Correct answer: b) 0.125 g. Q2: Selenium 83 has a half-life of 25.0 minutes.

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web each radioactive isotope will have its own unique half-life that is independent of any of these factors. Figure page index 1 for cobalt 60, which has a half-life of 5.27 years. 50% remains after 5.27 years; one half-life. 25% remains after 10.54 years; two half-lives. 12.5% remains after 15.81 years; three half-lives, and so on.

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