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web a key step in balancing chemical equations is correctly identifying the formulas of the reactants and products for example consider the reaction between calcium oxide  $\text{CaO}$  s and  $\text{H}_2\text{O}$  l to form aqueous calcium hydroxide b is it possible to balance the equation if you incorrectly identify the product as  $\text{CaOH}$  aq and if so what is the equation

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