

# 10.1 Periodic Trends



## 10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

### Section 10.1: Introduction

In your notebook, read aloud the modern periodic table.

Copy the lettered items in the first row on the left below from the periodic table to complete the table on the right.

_____
_____
_____
_____
_____

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

#### Column A

- \_\_\_\_\_ 10. It exists in the periodic table.
- \_\_\_\_\_ 11. It has no line periodic table.
- \_\_\_\_\_ 12. Group 1 elements.
- \_\_\_\_\_ 13. Elements that are strong oxidizing agents.
- \_\_\_\_\_ 14. Group 17 elements.

#### Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 15. There are two representative elements of elements.
- \_\_\_\_\_ 16. Most elements have lines in the periodic table.
- \_\_\_\_\_ 17. Group 1 elements except for hydrogen are metals in the solid state.
- \_\_\_\_\_ 18. Group 17 elements are the active metal.
- \_\_\_\_\_ 19. Group 1 elements are highly reactive elements in the periodic table.
- \_\_\_\_\_ 20. Group 17 elements are very reactive elements in the periodic table.
- \_\_\_\_\_ 21. Elements have properties of both metals and nonmetals.

# Neutralization Study Guide For Content Mastery

**C Cleary**



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## **Neutralization Study Guide For Content Mastery** Book Review: Unveiling the Power of Words

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