

Mystery Powder Lab!

Purpose: One of the employees at our company spilled some of our powders. We now have an unknown substance that we need to help identify. You will want to learn the properties of each white powder before trying to figure out the mystery combination!

Background: The shape, smell, color, size and how an item reacts are called properties. Some properties are physical and some are chemical. The color, shape, and size are physical properties of an object. Physical properties can be observed without changing the chemical property. Chemical properties include how an object behaves or reacts to something such as an acid. For us to use a chemical property, an item has to actually change chemically. Chemicals are the elements that study the properties of a substance. Both chemical and physical properties are used to identify an item.

In forensic science and chemistry, being able to identify and analyze unknown substances is very important. A police officer might find a white powder at a suspect's house, like the poison master's pot to find out if that white powder is flour, sugar, or a illegal drug. With the use of scientific problem solving, the mystery powder can almost always be correctly identified.

Materials:

Corn Starch	Mystery Powder	Magnifying Glass
Salt	Water	Reaction Plate
Plaster of Paris	Vinegar	Eyedropper
Baking Soda	Table Salt 10%	Stopper

Procedure:

- Use the spoon to put the same amount of corn starch in  both of the wells of your reaction plate.
- Observe the powder with your magnifying glass and record your observations in the data table below.

Process	Magnifying Glass Observations	Water	Vinegar	Stopper
Corn starch				
Salt				
Plaster of Paris				
Baking Soda				
Mystery Powder				

- Use the spoon to place 10 drops of water into well 1. Use with your toothpick and then observe how well the powder dissolved in the water. Record this in the data table.
- Gently squeeze 3 drops from the bottle of vinegar into well 2. Observe the color of your powder. Record your observations in the data table.

Vinegar is an **acid** or **base**, meaning that it changes if something has been in it. It will turn **colorless** if the powder doesn't have any starch, but will turn **blue-black** or **purple** if the powder has starch in it. If you have extra vinegar after this experiment, try heating some of the foods in your kitchen to see if they have starch in them.

- Gently squeeze 3 drops from the bottle of vinegar into well 3. Observe what happens and record your observations in the data table. Vinegar will react with bases (chemicals other an acid or a base, but special kinds of chemicals). Examples of acids are lemon juice, vinegar and coffee. Examples of bases are bleach, soap and baking soda. When an acid reacts with a base they react to each other to form a new substance.
- Carefully rinse and dry your stopper and reaction plate.
- Repeat steps 1-3 with the salt, plaster of paris, and baking soda.
- Now it is time to figure out your mystery powder! Follow steps 1-4 with it and then use its properties to figure out which two powders combined to make it.

What the mystery powder is a mixture of _____ and _____ because _____

Questions:

- What properties did all of the powders have in common?
- How can you tell the difference between Cornstarch and Plaster of Paris?
- How can you tell the difference between Salt and Baking Soda?
- How could properties help you tell the difference between each powder?
- What is a physical property? Give an example.
- What is a chemical property? Give an example.

Extensions:

Signs that a chemical change has taken place are a change in color, smell, or temperature. Gas bubbles can form and you might hear a sound. Did you see any of these signs when you were testing the mystery powder? Put a star in each box of the data table where a chemical change took place.

Mystery Powder Lab Answers

N Nodding

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Differentiating Science Instruction and Assessment for Learners With Special Needs, K-8 Kevin D. Finson, Christine K. Ormsbee, Mary M. Jensen, 2011-03-02 Field tested strategies for teaching science to students with special needs Teachers are required to provide appropriate science instruction to all students including children with special needs However they are often left on their own to figure out how to effectively differentiate lessons and activities Help is here This timely practical guidebook shows general and special educators how to retool science activities and assessments for students with learning disabilities behavior disorders and more The authors cover a broad range of topics in an orderly concise fashion including National and state requirements for student learning and science literacy Pedagogical strategies for collaborative learning groups self paced learning centers literature circles and team projects Grade appropriate ways to revise science activities and assessments for biology earth science and physical science lessons Step by step instructions for using rubrics for evaluation revision and assessment Information on teacher collaboration and specific disabilities Also included are vignettes and checklists to assist teachers in bridging the gap between science and special education instruction and assessment By adjusting the content teaching critical thinking and providing a variety of ways for learners to demonstrate their knowledge you will give all students the chance to achieve academic success in science

No Child Left Behind? Elizabeth Blake, 2008 From shootings gangs drugs teen pregnancies and even some administrators a science teacher at an inner city high school describes the challenges she faced as she tried to reach her at risk students She believes giving teachers and students the help they need not more tests is the only way to truly begin to leave no child behind

CSI Expert! Karen K. Schulz, 2021-09-03 Crime scene investigation is hotter than ever and kids everywhere will love learning about how their favorite detectives use science to figure out unsolvable thefts arsons mysteries and more CSI Expert Forensic Science for Kids includes more than 25 in depth activities on fingerprinting evidence collection blood stain identification forensic careers ballistics and much more The author of the best selling Crime Scene Detective series combines more than a decade of experience teaching forensic science to middle school students with the latest technology and research in criminal investigations in the intriguing standards based scientific study included in CSI Expert Students will love collecting dental impressions studying their classmates fingerprints looking at tool marks left at the scene of the crime analyzing mysterious powders and discovering the various types of counterfeit checks Each lesson includes a realistic case for students to crack using the knowledge they've learned about analyzing forensic evidence and the book also includes an assessment assignment that teachers can employ to test their students learning Both kids and teachers will be able to easily implement the book's hands on detailed and exciting forensic science experiments using everyday materials After completing these activities kids will be begging for more fun science learning Grades 5-8

Case Studies in Science Education and the Life Lab Science Program Michael Arenson, 1989

Cognitive Theory as the Basis for Design of Innovative Assessment

Robert Glaser,1992 **Teaching Handicapped Students Science** Marshall Corrick,Marshall E. Corrick,1981 One of five volumes intended to help teachers of mainstreamed handicapped children the book presents 19 papers on science instruction Papers address the issues of goals prerequisites approaches cooperative multisensory and concept analysis materials barriers and evaluation The following titles and authors are included Science for the Handicapped Can We Justify It A Scheinker C Coble Strategies in Science Instruction for Students with Special Learning Disabilities Prompted by Piaget s Formulations N Adibe Metric Measurement for Blind Students F Franks Mainstreaming Handicapped Students into Science Classrooms R Johnson D Johnson Cooperative Goal Structures and the Mainstreaming of Handicapped Students M Corrick Creativity for the Mentally Handicapped D Daugs An Audio Tactile Approach to Science Education for Visually Impaired Students R Harris Chemistry Experiments for the Deaf Secondary Student A Visual Approach R Menchel The Man the Child and the Flower Laboratory Science for the Handicapped Student Teaching Mainstream Strategies D Hadary Concept Analysis A Model for Teaching Basic Science Concepts to Intellectually Handicapped Students J Cole et al The Visually Impaired High School Student Can See Her or His Progress in the Regular Science Classroom M Ovnik Strategies for Stimulating Scientific Inquiry for All Students P Welliver Multisensory Science Education Meeting Special Challenges L Malone L De Lucchi Using Science to Strengthen Communication Skills of Hearing Impaired Students D Orlich K Black Science for the Developmentally Disabled J Trotta Science for Deaf Students Curriculum Suggestions for Grouped and Mainstreamed Programs J Stolte S Smith Some Thoughts on Teaching Science to the Mentally Handicapped Secondary Student M Mathias R Johnson Some Psychological Considerations in the Education of Blind Students E Gough and Criterion Referenced Testing and Prescriptive Instruction in the Science Classroom H Lang CL [Handbook of Student Engagement Interventions](#) Jennifer A.

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on increasing engagement and school completion for all students Emphasizes multidimensional approaches to disengagement **The Saint Mystery Magazine** ,1964 **El-Hi Textbooks and Serials in Print, 1996** R R Bowker Publishing, Bowker, 1996-02 **AIM Report** Accuracy in Media, inc,1995 **Addison-Wesley Introduction to Physical Science** Michael B. Leyden,1988 *Treasury of Literature* Roger C. Farr,1995 Prentice Hall Literature Prentice Hall (School Division),2001 **Machine Design** ,1994 Voices and Reflections Harcourt School Publishers Staff,Roger C. Farr,1995 **Readers' Guide to Periodical Literature** ,1941 An author subject index to selected general interest periodicals of reference value in libraries The Bookseller ,1956 **Lab World** ,1979 **Science News-letter** ,1965

Discover ,2004-07

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