

SCIENCE TEST – 2
TOPICS : ATOMS & MOLECULES

Class IX	Time 1.5 HRS
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General Instructions :

- 1) All questions are compulsory.
- 2) Each question carry EQUAL marks.

1. State the law of constant proportion. The percentage of Ca, C and O in a sample of CaCO_3 is found to be Ca (40 %), C (12 %) & O (48 %). If the law of constant proportion is true, what weight of these elements will be present in 1.5 g of sample of this compound?
2. Give three important limitations of Dalton's atomic theory. Which are the two postulates of Dalton which form basis of two laws of chemical combination?
3. Answer the followings ---
 - a) Unit for measuring size of atom.
 - b) Symbol used by Dalton for Hydrogen.
 - c) Name of elements represented by Au and Na.
 - d) What is the difference between Co and CO.
 - e) Why was Oxygen used as the standard element for Relative Atomic Mass.
 - f) What is atomicity of an element? Name a mono atomic and tetra atomic element.
 - g) Give one example each of positive polyatomic & negative polyatomic ion.
4. What is meant by valance of an element ? Give three important rules for writing chemical formulae. Write the formulae of Calcium bicarbonate , Ammonium phosphate , potassium chromate and Copper nitrate.
5. Name the following compounds ---
 - a) $\text{Fe}_2(\text{SO}_4)_3$
 - b) $(\text{NH}_4)_2\text{S}$
 - c) FePO_4
 - d) NaNO_3
 - e) K_2SO_4
 - f) $\text{Al}_2(\text{SO}_4)_3$
6. An element " X " forms an oxide with formula X_2O_3

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