

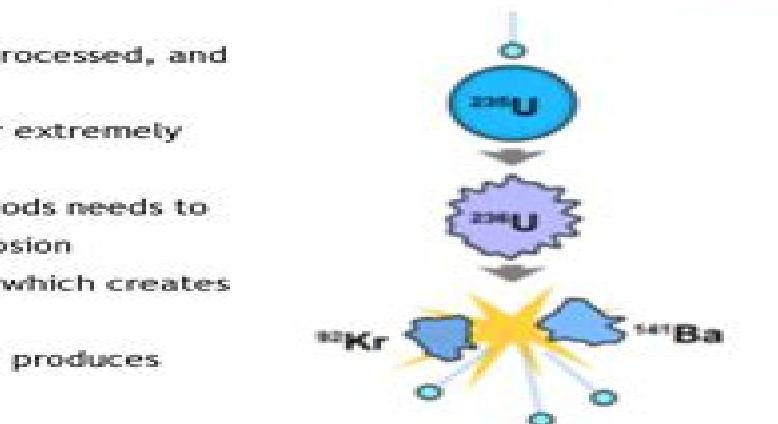
Nuclear Power

Nuclear Energy

- When the nucleus of an atom is split, it releases a huge amount of energy called **nuclear energy**
 - Scientists and engineers have learned to split nuclei and to control the release of energy
- Nuclear Energy Use**
 - Nuclear power plants use uranium, which is mined, processed, and then concentrated into fuel rods
 - When uranium atoms in the fuel rods are hit by other extremely tiny particles, they split apart
 - The number of tiny particles allowed to hit the fuel rods needs to be controlled, or they would cause a dangerous explosion
 - The energy from a nuclear power plant heats water, which creates steam and causes a turbine to spin
 - The spinning turbine turns a generator, which in turn produces electricity
- Consequences of Nuclear Power:** Nuclear power is clean and does not pollute the air, but the use of nuclear energy does create other environmental problems
 - Process of splitting atoms creates radioactive waste, which remains dangerous for thousands or hundreds of thousands of years
 - As yet, there is no long-term solution for storing this waste
- The development of nuclear power plants has been on hold for three decades because of catastrophic accidents (e.g. Three Mile Island and Chernobyl)
- Nuclear power appeared to be making a comeback as society looked for alternatives to fossil fuels
 - However, the 2011 disaster at the Fukushima Daiichi Nuclear Power Plant in Japan may have resulted in a new fear of nuclear power
- Nuclear power is a controversial subject and the future of nuclear power is murky

Connections

Nuclear fusion, or the conjoining of two nuclei, can also harness a great deal of energy and is considered another form of nuclear power.



When struck by a tiny particle, Uranium-235 breaks apart and releases energy.



Nuclear power plants like this one provide France with almost 80% of its electricity.

Nuclear Unit Study Guide

William Kenneth Metcalf

Nuclear Unit Study Guide:

Resources in Education, 1984 **Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e** Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry [Study Guide to Accompany Raven and Johnson Biology](#) David A. Stetler, 1986 **Study Guide with Programed Units for Hilgard, Atkinson, and Atkinson's Introduction to Psychology, Fifth Ed** Rita L. Atkinson, Richard C. Atkinson, 1971 **Laboratory Assistants Examination Review Book** Alvin M. Ring, 1971

Reading Guide University of Virginia. School of Law, 1970 *Fundamentals of Chemistry, Study Guide* James E. Brady, John R. Holum, 1988-04-20 This Third Edition of the widely used fundamentals textbook for science majors maintains the conversational writing style that made the previous editions so popular while including up to date treatments of important and current topics Emphasizes descriptive chemistry chemical reactions and properties while maintaining a solid treatment of chemical principles Common chemicals are used whenever possible as examples in both theoretical discussions and in problems and exercises Incorporates many pedagogical aids each chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs

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