

28.2**NUCLEAR TRANSFORMATIONS**
SECTION REVIEW**Objectives**

- Use half-life information to determine the amount of a radioactive isotope left after a given time.
- Given a series of衰变 (radioactive decay) equations, determine the half-life for each isotope.

Key Terms

- isotopic stability
- radioactive decay
- positron
- beta-minus decay

Part A: Completion

After beta-minus decay, an element's mass number decreases by one, and its atomic number increases by one. Radioactive decay can be categorized into three types: alpha, beta-minus, and gamma.

What term describes an element that undergoes spontaneous radioactive decay? _____ 1. _____

What term describes the energy released during beta-minus decay? _____ 2. _____

What term describes an isotope that has a positive charge? _____ 3. _____

What term describes an isotope that has a negative charge? _____ 4. _____

What term describes decay that is characterized by _____ 5. _____

What term describes decay that is characterized by _____ 6. _____

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Nuclear Transformations Section Review Key

Mike Jess

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