

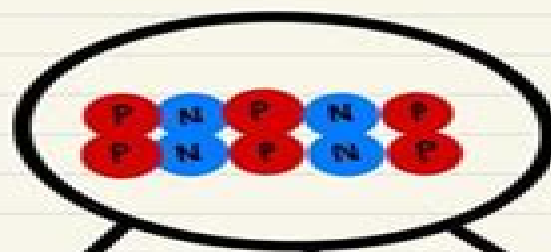
Nuclear Physics



- The number of protons defines an element.
- In an isotope, the number of neutrons is different.
- This means that the nucleon number is different.

- Isotopes with an unbalanced no. of protons and neutrons are more likely to be unstable.

- All radiation is ionising.
- It can remove electrons from atoms (e.g. DNA)
- This causes mutation and can lead to cancer



Alpha (α)



- Identical to Helium nucleus. (2P, 2N)
- Can be stopped by thin sheet of paper/few cm of air
- Very ionising, does most damage under skin.

Beta (β)



- Fast, high energy electron.
- Few mm of aluminium, 1m of air.
- Medium

Gamma (γ)



- High energy EM wave
- Several cm of lead or very thick concrete
- Low ionisation - easily passes through body

Background Radiation

- Naturally occurring e.g. rocks, soil, space (62%)
- Artificial e.g. medical, nuclear power, food + drink & weapons testing.

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Table of Contents Nuclear Physics 11 Notes

1. Understanding the eBook Nuclear Physics 11 Notes
 - The Rise of Digital Reading Nuclear Physics 11 Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying Nuclear Physics 11 Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nuclear Physics 11 Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nuclear Physics 11 Notes
 - Personalized Recommendations

- Nuclear Physics 11 Notes User Reviews and Ratings
- Nuclear Physics 11 Notes and Bestseller Lists
- 5. Accessing Nuclear Physics 11 Notes Free and Paid eBooks
 - Nuclear Physics 11 Notes Public Domain eBooks
 - Nuclear Physics 11 Notes eBook Subscription Services
 - Nuclear Physics 11 Notes Budget-Friendly Options
- 6. Navigating Nuclear Physics 11 Notes eBook Formats
 - ePub, PDF, MOBI, and More
 - Nuclear Physics 11 Notes Compatibility with Devices
 - Nuclear Physics 11 Notes Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nuclear Physics 11 Notes
 - Highlighting and Note-Taking Nuclear Physics 11 Notes
 - Interactive Elements Nuclear Physics 11 Notes
- 8. Staying Engaged with Nuclear Physics 11 Notes
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nuclear Physics 11 Notes
- 9. Balancing eBooks and Physical Books Nuclear Physics 11 Notes
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nuclear Physics 11 Notes
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Nuclear Physics 11 Notes
 - Setting Reading Goals Nuclear Physics 11 Notes
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nuclear Physics 11 Notes
 - Fact-Checking eBook Content of Nuclear Physics 11 Notes

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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