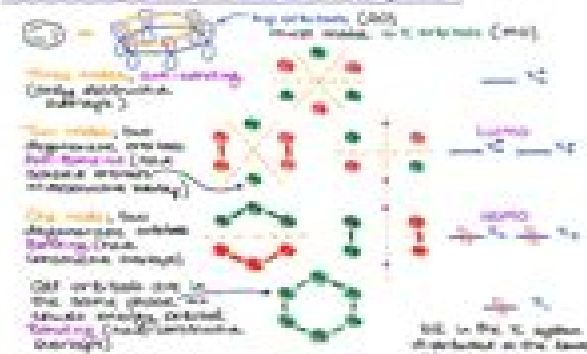
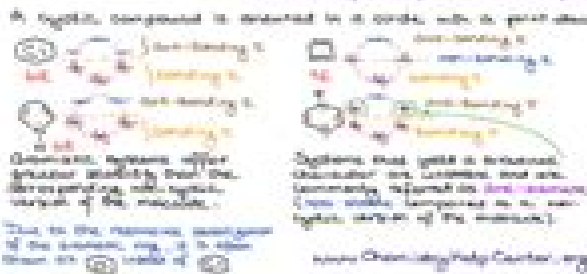


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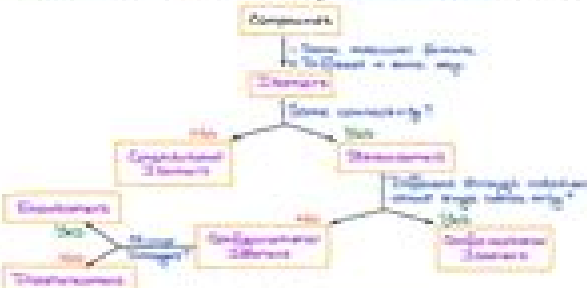


Free Radical Substitution - only way to produce substituted aromatic compounds for simple aromatic compound



Due to the resonance, the positive charge of the sigma complex is delocalized over the ortho and para positions.

- Summary of Stereochemical Relationships**
- Constitutional Isomers** - completely different molecules
- Conformational Isomers** - same molecule that only differ in 3D shape due to free rotation around single bonds
- Enantiomers** - non-superimposable mirror images that share physical & chemical properties in achiral environment
- Diastereomers** - non-superimposable, non-mirror images that have different physical & chemical properties
- Meso Compounds** - non-chiral due to the plane of symmetry, however, have chiral centers



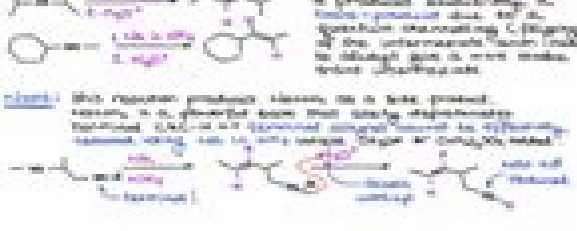
Stereochemical Relationship of Enantiomers

Enantiomers are mirror images, non-superimposable, and have identical physical properties in achiral environment, they are the only pair of molecules that are mirror images of each other. They are separated by chiral resolution, which involves separating them using physical means & converting them into original molecules.

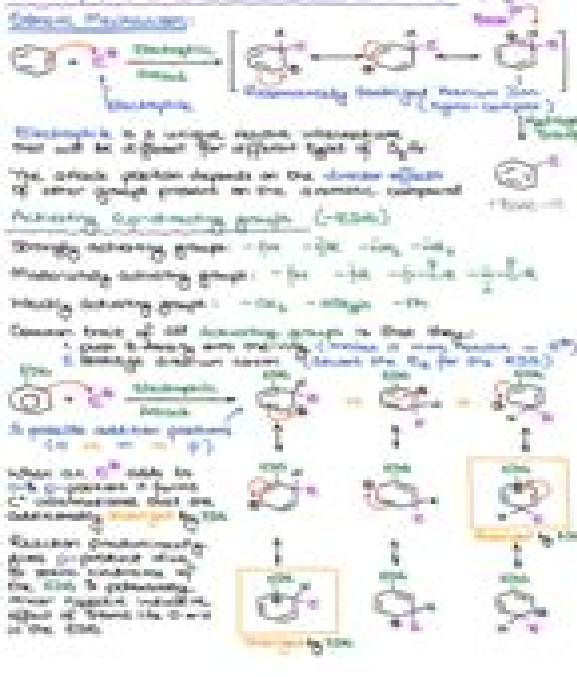
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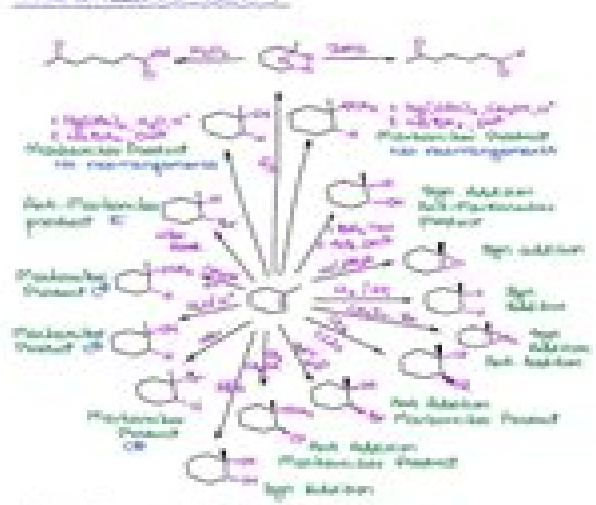
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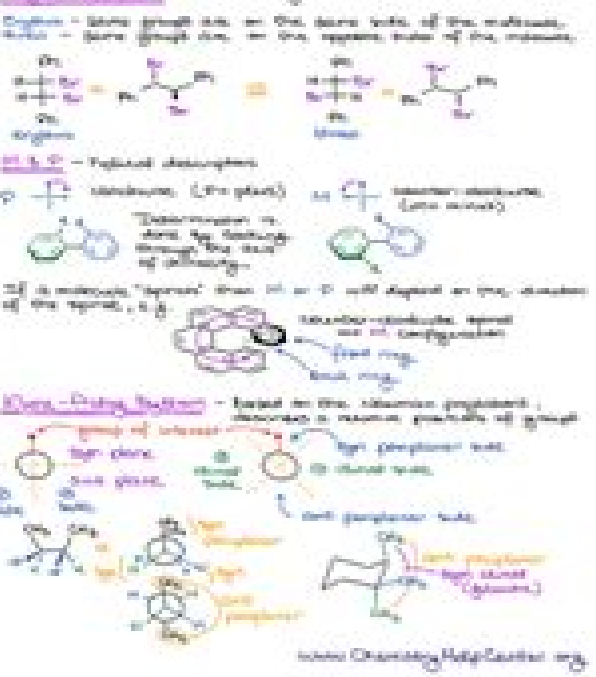
Aromatic Substitution Mechanism



Aromatic Substitution Reactions



Reaction of Nitrobenzene with Aromatic Substitution



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and bonds as they read the text and make the logic of organic chemistry clear and easily understood Each chapter ends with a list of frequently asked questions and answers followed by additional practice problems Answers are included in the Appendix

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