

3. Graphs:

concn - x axis = diff. concentrations  
 y axis = time  
 0 - at the end  $\rightarrow$  fastest

slow  
 steady  
 fast  
 slow

pH - x axis = pH  
 y axis = time

slow

temp - x axis = temp  
 y axis = time

discussion 18s - use data from  
 data table she  
 gave us

Discussion

4.3A - pg. 100 # 1, 2

4.3B - pg. 103 ~~work~~

no # question  $\rightarrow$  disc. # is not  
 numbered

4.3C - pg. 107 # 1, 2

Conclusion

Summarize the results.

Explain how changes in  
 environment affect  
 reaction rate of an enzyme.  
 - Be specific - concn, pH,  
 temp

2. Prepare the experimental test tubes (test-tube set II):
  - a. Number a second set (set II) of five standard test tubes 1 through 5. Beginning with tube 5 of set I, transfer 2 mL of this enzyme dilution into tube 5 of set II. Use a 5-mL pipette for the transfer. Rinse the pipette in distilled water, and repeat the procedure for tubes 4, 3, 2, and 1, transferring 2 mL of tube 4 (set I) into tube 4 (set II), and so forth. After these transfers have been carried out, test-tube set I will no longer be used and can be set aside.
  - b. Add 2 mL of pH 6.8 buffer solution to each of the tubes in the second set. Mix by rolling the tubes between your hands. Set these tubes aside.
  - c. Add 1 or 2 drops of I<sub>2</sub>KI to each compartment of four rows of a test plate. You will use a separate row for each concentration of amylase.
  - d. Using the second set of tubes, proceed with the tests beginning with tube 5.
    - (1) Using a clean 1-mL pipette, add 1 mL of the 1% starch solution to tube 5 and mix by rolling the tube between your hands. One team member should immediately record the time. This is time 0.
    - (2) Quickly remove 1 drop of the mixture with a disposable Pasteur pipette, and add it to a drop of I<sub>2</sub>KI in the first compartment on the test plate (time 0).



Remember, when the enzyme and substrate are together, the reaction has begun!

3. Sample the reaction mixture at 10-second intervals, each time using a new compartment of the test plate. Continue until a blue color is no longer produced and the I<sub>2</sub>KI solution remains yellow-amber (indicating the digestion of all the starch). Record the time required for the digestion of the starch in Table 4.6.
4. Repeat steps 1 through 4 for the other four concentrations (tubes 4, 3, 2, and 1 of set II).
5. Finish recording your findings in the Results section, and state your conclusions in the Discussion section.

**Results**

1. Complete Table 4.6 as you determine rates of digestion (time of starch disappearance) in different enzyme concentrations.

**Section 5.1A prelab**

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**Minjie Lin**

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newton s laws of motion in this chapter we continue to discuss these strategies and apply a step by step process

5 5 newton s third law university physics volume 1 openstax - May 12 2023

web state newton s third law of motion identify the action and reaction forces in different situations apply newton s third law to define systems and solve problems of motion we have thus far considered force as a push or a pull however if you think about it you realize that no push or pull ever occurs by itself