

# Biochemistry Study Guide

1. Organic molecules
  1. **Macromolecules:** Large complex molecules found only in living organisms.
  2. Define the term organic molecule. **Molecules that ave carbon at the center**
  3. Organic molecules are organized around central carbon skeleton.
2. Hydrocarbons
  1. Carbon atoms have four electron in their valence shell, so they can form up to four covalent bonds with other atoms.
  2. Define the term hydro-carbon. **Organic molecules consisting entirely of hydrogen and carbon**
  3. Why are hydrocarbons hydrophobic and insoluble in water? **Hydrocarbons are non polar due to slightly differing electronegativities between carbon and hydrogen**
  4. Describe the difference between aliphatic hydrocarbons and aromatic hydrocarbons. Be able to recognize the structural formula of each. **Aliphatic contain a conjugated bond system, while aromatic do not**
3. Organic molecules are important for living organisms, because of their complex structure.
  1. Organic molecules can form isomers. Define the term isomer. **A molecule that hase the same molecular formula but in different structural formulas**
  2. Organic molecules can form covalent bonds with functional groups. Define the term functional group. **A group of atoms that are attached to the carbon skeleton of an organic molecule**
  3. Be able to name the seven functional groups listed in Figure 2.27 on p. 53, and to recognize their structure.
  4. Be able to recognize each of the seven functional groups within a larger organic molecule.
  5. Macromolecules can be very large.
  6. Name the four groups of macromolecules. **Carbohydrates, protiens, nucleic acids, and lipids**
  7. Carbohydrates, proteins, and nucleic acids can be very large, because they are **polymers** that are built from monomers.
  8. Monomers are linked together during a **dehydration synthesis**. A water molecule is always a product of a dehydration synthesis.

# Organic And Biochemistry Study Guide

**Peter C. Scott, Morris Hein**



## Organic And Biochemistry Study Guide:

**Introduction to Organic and Biochemistry** Peter C. Scott, Morris Hein, 1993-02 Designed for professors who prefer to teach general chemistry topics from one text and organic and biochemistry topics from another this text offers step by step and easy to understand coverage of the important functional groups reactions and macromolecules that are essential for health science students A dynamic full color presentation and numerous applications add to the quality of the presentation Content corresponds to Chapter One and Chapters 21-37 of College Chemistry An Introduction to General Organic and Biochemistry Fifth Edition by the same authors Clarity meticulous accuracy and a step by step approach that students can and do understand have become hallmarks of the Hein authorship This new text is no exception Anticipating student problems before they occur the authors move at a manageable pace offering carefully worked out examples with alternate methods of solution practice problems with answers review of concepts review of key terms and a number of other learning aids to ensure student mastery of important material *General, Organic, and Biochemistry Study Guide* Marcia L. Gillette, Ira Blei, Wendy Gloffke, George Odian, 2006-01-27 This study guide provides reader friendly reinforcement of the concepts covered in the textbook Features include Chapter outlines Are you able to Worked text problems Fill ins Test yourself Concept maps Can also be used for Blei and Odian's Organic and Biochemistry *General, Organic, and Biochemistry, Study Guide* William Henry Brown, 1983 **Organic and Biological Chemistry, Study Guide** John R. Holum, 1996-01-18 Explains recent advances in environmental studies and the molecular basis of life Designed for those in the health care field it focuses solely on organic and biochemistry General, Organic and Biological Chemistry, Student Study Guide and Solutions Manual Kenneth W. Raymond, 2008-01-14 This General Organic and Biochemistry text has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry Students need have no previous background in chemistry but should possess basic math skills Raymond was crafted to take advantage of recent trends in the GOB market It is a shorter lighter book with a new integrated table of contents that develops general organic and biochemistry topics together rather than in isolation In introducing GOB material this text uses an integrated approach in which related general chemistry organic chemistry and biochemistry topics are presented in adjacent chapters This approach helps students see the strong connections that exist between these three branches of chemistry and allows instructors to discuss these interrelationships while the material is still fresh in students minds This integration involves the following sets of chapters Chapter 3 Compounds and Chapter 4 An Introduction to Organic Compounds An introduction to bonding and compounds is followed by a look at the members of a few key organic families Chapters 3, 4 and 6 Reactions A study of inorganic and organic compounds is followed after a look at gases liquids and solids in Chapter 5 by an introduction to their reactions Chapter 7 Solutions and Chapter 8 Lipids and

Membranes A discussion of solubility is followed by a look at the importance of solubility in biochemistry Some reactions from Chapter 6 are reintroduced Chapter 9 Acids and Bases and Chapter 10 Carboxylic Acids Phenols and Amines Principles of acid base Chemistry from an inorganic perspective are followed by a chapter on the organic and biochemical aspects of this topic Chapter 11 Alcohols Aldehydes and Ketones and Chapter 12 Carbohydrates An introduction to the chemistry of alcohols aldehydes and ketones is followed by a presentation of related biochemical applications **Study Guide for**

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