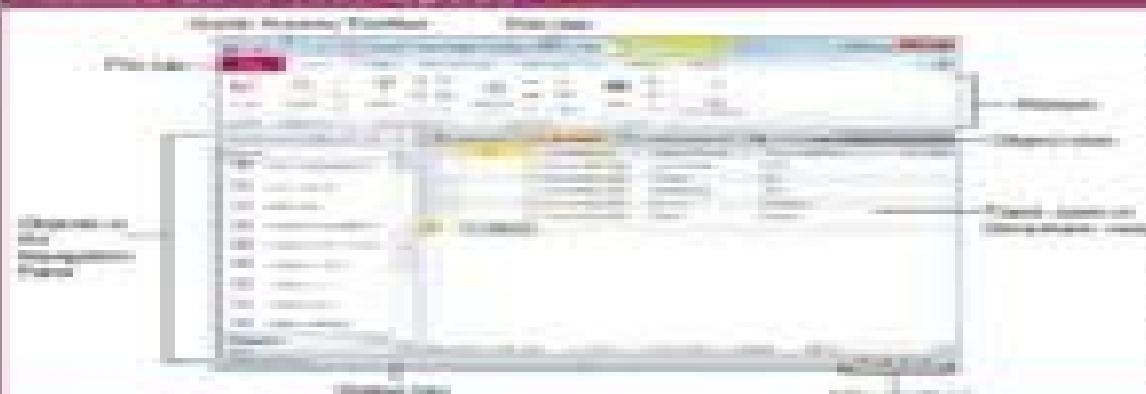


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- 1000 **Polymers** are long chain molecules of many repeating units called monomers.
  - 1001 **Classifying** them into naturally-occurring (e.g. starch) and synthetic (e.g. polythene) polymers.
  - 1002 **Explaining** the chemical structure (the formulae) of some ways in which small molecules can be linked.
  - 1003 **Explaining** common types (e.g. of fibres or plastics) in common usage.
  - 1004 **Explaining** common chemical reactions and starting with the polymerisation of ethene to give polythene.
  - 1005 **Remember** some groups of monomers called **functional groups** which are common in polymers.
- The **Monomers** are **Alkenes** (Double bond) the subject is the **Functional group**.
  - The **Monomers** are **Alkenes** (Double bond) the **Functional group** is the **Double bond** and the **Monomers** are **Alkenes** (Double bond) and the **Monomers** are **Alkenes** (Double bond).
  - The **Monomers** are **Alkenes** (Double bond) and the **Monomers** are **Alkenes** (Double bond).

- [illegible]

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Cost	Cost = 10
Energy	Cost = 12
Protein	Cost = 10
Fruit	Cost = 9
Vegetables	Cost = 10
Grains and	Cost = 10

100

Phosphorus	200-400 mg
Chlorine (chloride)	1000-1500 mg
Sulfur (sulfate)	1000-1500 mg
Sodium (chloride)	1000-1500 mg

1. *Journal of the American Medical Association*, 1997; 277: 1001-1005.

Alcohol	0.0000
Alkaloids	0.0000
Carbohydrates	0.0000

100

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# Microsoft Access 2010 Quick Check

**Carole Matthews, Marty  
Matthews, Bobbi Sandberg**



## **Microsoft Access 2010 Quick Check:**

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