



# Micros Workstation 5 Manual

**Xiang Xie**

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web general information cylinder block the 2rz fe has a cast iron cylinder block a 95.0 mm 3.74 in cylinder bore and 86.0 mm 3.39 in piston stroke give the 2rz fe engine a total of 2 438 cc 148.8 cu in of displacement compression ratio rating is 9.5:1

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web jan 25 2020 bore is 86 mm and stroke is 86 mm the 1rz e is the fuel injected version of the 1rz with a 9.0 to 1 compression ratio output is 101 108 hp at 5 400 rpm with 118 123 lb ft 161 167 n m of torque at 2 800 rpm the 2rz is a 2 4 l 2 438 cc version

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web with a strong acid and strong base titration strong acids and strong bases refer to species that completely dissociate to form ions in solution by contrast weak acids and bases ionize only partially and the ionization reaction is reversible

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3 655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1 1 we could experimentally determine the number of moles of the acid in the solution to be 000731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the  $K_a$  or  $K_b$  from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $K_{a1}$  and  $K_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1 1  
hcl aq naoh aq nacl aq h<sub>2</sub>o 1 hcl a q naoh a q nacl a q h 2 o l one mole of hcl hcl would be fully neutralized by one mole of naoh naoh

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide naoh and oxalic acid h<sub>2</sub>c<sub>2</sub>o<sub>4</sub> 2h<sub>2</sub>o and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web acid base reactions can be used to determine concentrations of an acid or a base using stoichiometric calculations and titrimetric analysis the volume of a solution required to react

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pka of an unknown acid or pkb of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance h<sub>2</sub>c<sub>2</sub>o aq 2naoh aq na<sub>2</sub>c<sub>2</sub>o<sub>4</sub> 4h<sub>2</sub>o to differentiate an acid from a base in a solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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