

Organic Chemistry II Study Guide for Exam I
Chapters 10, 11, and 14

- Classification of Alcohols
 - Alcohols are classified according to how many carbons are attached to the carbon containing the -OH group.

- Methyl, primary, secondary, and tertiary

- Nomenclature of Alcohols

- Naming an alcohol

- Start numbering the carbon chain to reach the -OH group first
 - Include the location (carbon number) of the -OH group



- 2-methylhexan-2-ol

- Naming an alcohol as a substituent

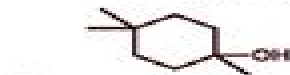
- Do this when alcohol is NOT the highest priority group (carboxylic acid, etc.)
 - Start numbering the carbon chain at the higher priority group
 - Use the prefix hydroxy- and include the location (carbon number) of the -OH group



- 4-hydroxybutanoic acid

- Naming cyclic alcohols

- If the -OH is the highest priority group, do NOT include the location (carbon number)
 - It is implied so it is unnecessary



- 1,4,4-trimethylcyclohexanol

- Naming diols

- When 2 -OH groups are present and is the highest priority group, use the suffix -diol
 - Include the location (carbon number) of BOTH -OH groups



- Propane-1,3-diol

Organic Study Guide

**T. W. Graham Solomons, Craig B.
Fryhle, Scott A. Snyder**



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Study Guide and Solutions Manual for Essential Organic Chemistry Paula Yurkanis Bruice, 2015-03-04 This Study Guide Solution Manual contains learning objectives chapter summaries and outlines as well as examples self tests and concept questions as well as complete step by step solutions to selected problems

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