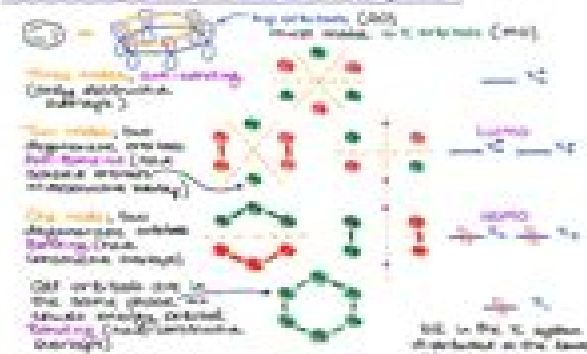


Mechanism of Reaction in Aromatic Compounds



Final Cycle Diagram - shows the overall mechanism of electrophilic aromatic substitution.

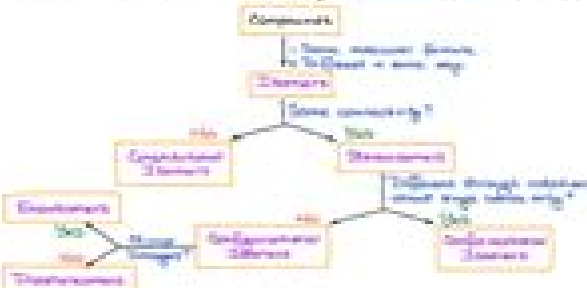
A cyclic compound is considered to be a cyclic, with a great deal of resonance.



Due to the resonance, the average bond length in benzene is 1.39 Å, which is intermediate between a single and a double bond.

Summary of Stereochemical Relationships

- Constitutional Isomers** - completely different molecules
- Conformational Isomers** - same molecule that only differ in 3D shape due to free rotation around single bonds
- Enantiomers** - non-superimposable mirror images that share physical & chemical properties in achiral environment
- Diastereomers** - non-superimposable, non-mirror images that have different physical & chemical properties
- Meso Compounds** - non-chiral due to the plane of symmetry, however, have chiral centers



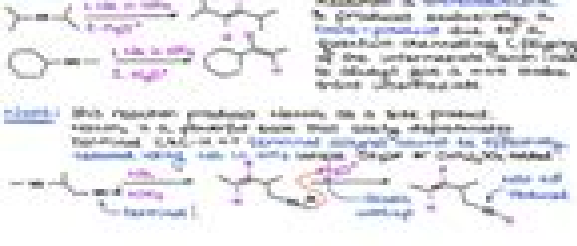
Stereochemical Relationships of Enantiomers

Enantiomers are mirror images of each other. They have the same physical & chemical properties in an achiral environment, but they differ in their interaction with plane-polarized light (optical activity) and in their reaction with chiral reagents.

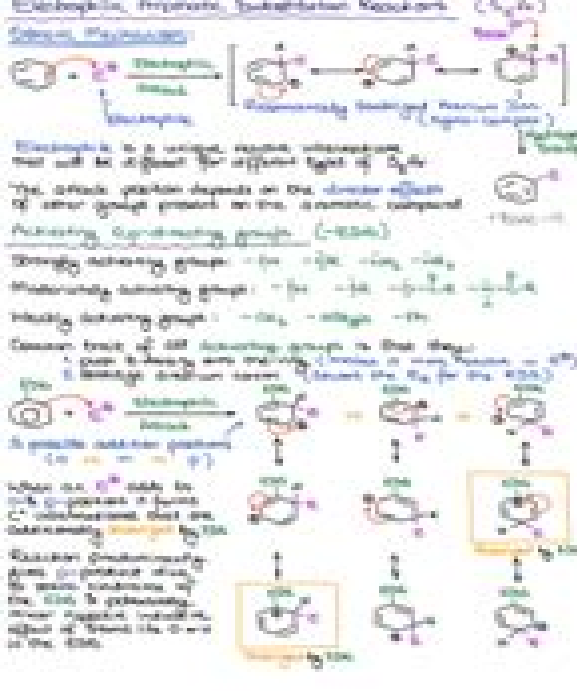
Reaction of Alkylated with Nitration in Liquid Ammonia



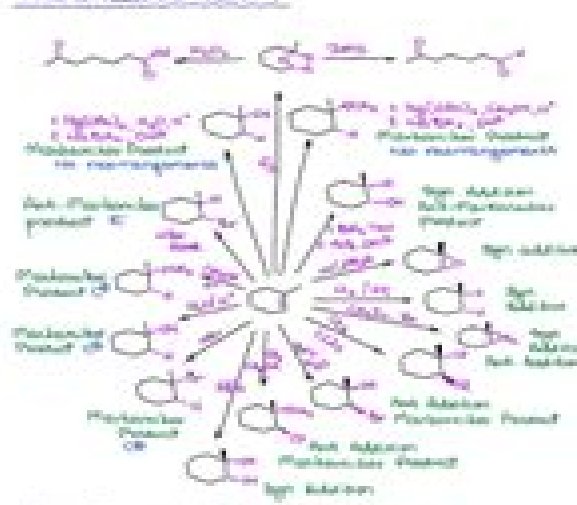
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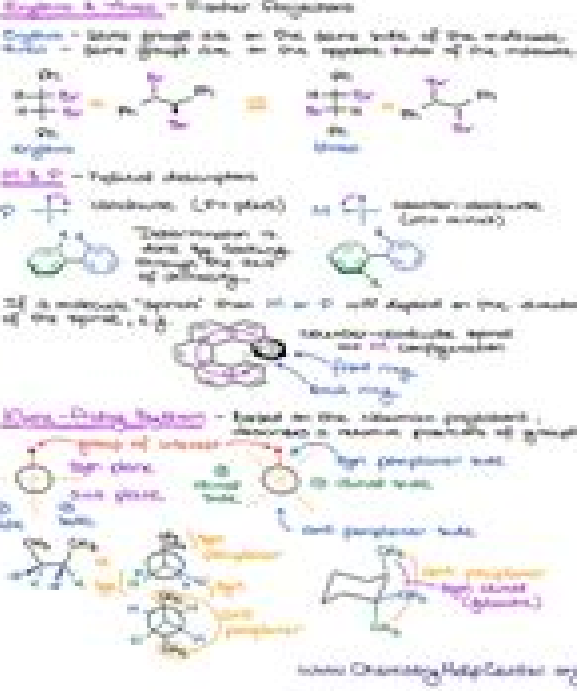
Alkene Reaction Mechanism



Alkene Reaction Mechanism



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Organic Chemistry Guide

Paula Yurkanis Bruice



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