



Ipad This Month

Guy Hart-Davis



IPad This Month:

IPad 2 J.D. Biersdorfer, 2011-11-16 Looks at the latest features and functions of the iPad2 covering such topics as accessing the Internet setting up an email account using built in apps playing games using iTunes and storing data content in iCloud

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web elements in the second row of the periodic table place their electrons in the $2n$ shell as well as the $1n$ shell for instance lithium li has three electrons two fill the $1s$ orbital and the third is placed in the $2s$ orbital giving an electron configuration

of 1 s 2 2 s 1

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web electron configurations and trends answer these discussion questions given the following part of an electron configuration 3s² explain which part refers to the number of electrons the energy level and the sublevel what is the maximum number of electrons that can occupy a single orbital state and explain hund s rule

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web the electron removed during the ionization of beryllium he 2s² 2 is an s electron whereas the electron removed during the ionization of boron he 2s² 2p¹ is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium 1s² ne 10 neon 1s² 2s² 2p⁶

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is 1s² 2s² 2p⁶ 3s² 3p⁶ 4s¹ or simply put ar 4s¹ q2 what is the total number of electrons that can be filled in s p and d subshells

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