



Balancing Oxidation-Reduction Reactions

- *Law of conservation of mass*: the amount of each element present at the beginning of the reaction must be present at the end.
- *Conservation of charge*: electrons are not lost in a chemical reaction.

Half Reactions

- Half-reactions are a convenient way of separating oxidation and reduction reactions.

Oxidation Reduction Reaction Prentice Hall Answers

M Lipman



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