

1. Please give an answer for the following question in Celsius. The problem requires:

$$10^3 \text{ J} = 10^3 \text{ W} = 1000 \text{ W}$$

The particle represented by Δ^0 is:

A) He^+ B) He^- C) He^0 D) He^2

2. Which isotopes will spontaneously decay and emit neutrinos with a charge of $-1/2$?

A) Li^7 B) Li^6 C) Li^8 D) Li^9

3. Which of these particles has the greatest mass?

A) alpha B) beta
C) neutrino D) gamma

4. Please give an answer for the following question in Celsius. The problem requires:

$$\text{Proton} \rightarrow \Delta \rightarrow \text{He}^+$$

The particle is represented by Δ :

A) alpha B) beta
C) neutrino D) gamma

5. Alpha particles will form gamma rays when the:

A) mass, only
B) charge, only
C) both mass, and charge
D) neither mass nor charge

6. Which statement best describes gamma radiation?

A) It has a mass of $1/2$ and a charge of 0 .
B) It has a mass of 0 and a charge of 0 .
C) It has a mass of 0 and a charge of 0 .
D) It has a mass of 0 and a charge of -0.2 .

7. Which type of radiation is most similar to single-energy X-rays?

A) alpha B) beta
C) neutrino D) gamma

8. Please use the initial masses of atoms that you gave. You are only 25,000 grams of an original 1,000 g-given sample of C^{14} remaining after 5000 years?

A) 238.0 g B) 50.0 g
C) 11.4 g D) 17.1 g

9. Which isotopes are most commonly used for the detection of cancerous tumors?

A) Fe^{55} B) Fe^{59} C) Fe^{60} D) Fe^{62}

10. Look at Reference Table 3. What fraction of a radioactive ^{100}Sr sample would remain unchanged after 5000 years?

A) $\frac{1}{2}$ B) $\frac{1}{4}$ C) $\frac{1}{8}$ D) $\frac{1}{16}$

11. Which is the healthiest used decay mode of $\text{Ra}-226$?

A) $\text{Ra}-226$ alpha used alpha decay
B) $\text{Ra}-226$ alpha used beta decay
C) $\text{Ra}-226$ alpha used alpha decay
D) $\text{Ra}-226$ alpha used beta decay

12. After 100 days, $1/2$ milligrams of an Ra -methylate sample will be measured for its Ra remaining unchanged. Which is the half-life of this element?

A) 5 days B) 12 days
C) 100 days D) 400 days

13. Which reaction converts one atom of one element to one atom of another element?

A) ionization B) polymerization
C) isotope fission D) isomerization

14. Radiation systems will be used in radiation therapy treatment. Cobalt-60 undergoes beta decay. What type of radiation reactions is used?

A) nuclear fission
B) nuclear fission
C) nuclear fusion
D) nuclear fusion

15. In which reactions is energy converted to energy by the presence of "fission"?

A) $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$
B) $\text{U}^{235} + \text{fiss} \rightarrow \text{Th}^{231} + \text{He} + \text{2 fission}$
C) $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$
D) $\text{U}^{235} \rightarrow \text{U}^{231}$

16. Which statement best shows the relationship in a fission reaction?

A) Energy emitted splits into lighter nuclei.
B) Light emitted from fission is heavier nuclei.
C) Energy is released and has visible differences in mass.
D) Energy is absorbed and mass is added after mass is released.

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