

1. Please give an answer for the following question in Celsius. The problem requires:

$$100^\circ F = 37.8^\circ C + 273^\circ$$

The problem is represented by $37.8^\circ C + 273^\circ$ in:

A) Celsius B) Fahrenheit C) Kelvin D) Rankine

2. Which isotopes will spontaneously decay and emit radiation with a charge of $+1.27^\circ$?

A) ^{100}Ru B) $^{100}\text{Ru}^+$ C) $^{100}\text{Ru}^-$ D) $^{100}\text{Ru}^{2+}$

3. Which of these particles has the greatest mass?

A) alpha B) beta C) neutrino D) gamma

4. Please give an answer for the following question in Celsius. The problem requires:

$$100^\circ C = X + 273^\circ$$

The particle is represented by X :

A) alpha B) beta C) neutrino D) gamma

5. Alpha particles will have greater loss of mass than:

A) muons, only B) electrons, only C) beta muons, and muons D) neither muons nor electrons

6. Which statement best describes gamma radiation?

A) It has a mass of 1.67×10^{-28} charge of 0 .
 B) It has a mass of 0 and a charge of 0 .
 C) It has a mass of 0 and a charge of 0 .
 D) It has a mass of 0 and a charge of $+0.2$.

7. Which type of radiation is most similar to single-energy X-rays?

A) alpha B) beta C) neutrino D) gamma

8. Please use the initial masses of atoms that you gave. You are only 25,000 atoms of an isotope of ^{100}Ru (5-g atoms sample of $\text{C} \times 14$ isotope) when you:

A) 2.00×10^{-23} g B) 5.00×10^{-23} g
 C) 1.1×10^{-23} g D) 1.7×10^{-23} g

9. Which isotopes are most commonly used for the detection of the presence of cancerous tissue?

A) ^{131}I B) $^{131}\text{I}^-$ C) $^{131}\text{I}^+$ D) $^{131}\text{I}^{2+}$

10. Look at Reference Table 3. What fraction of a radioactive ^{100}Ru sample would remain unchanged after 50.0 years?

A) $\frac{1}{2}$ B) $\frac{1}{4}$ C) $\frac{1}{8}$ D) $\frac{1}{16}$

11. Which is the healthiest and safest mode of $\text{Ra}-226$?

A) ^{226}Ra alpha and alpha decay
 B) ^{226}Ra alpha and beta decay
 C) ^{226}Ra alpha and gamma decay
 D) ^{226}Ra alpha and beta decay

12. After 100 days, $\frac{1}{2}$ milligrams of an $\text{Ra}-226$ sample will be replaced by a new radioactive isotope unchanged. Which is the Ra-226 of this element?

A) 8 days B) 12 days
 C) 16 days D) 4 days

13. Which reaction converts one atom of one element to one atom of another element?

A) ionization B) polymerization
 C) isotope fission D) isomerization

14. Radiation systems will be used in radiation therapy treatment. Could an undergo beta decay? This type of radiation reactions is called:

A) natural ionization
 B) artificial ion ionization
 C) nuclear fission
 D) nuclear fusion

15. In which reactions is energy converted to energy by the presence of "fission"?

A) $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$
 B) $\text{U-235} \rightarrow \text{Ba} + \text{Kr} + \text{He} + 3\text{Ne}$
 C) $\text{U-238} \rightarrow \text{Ba-234} + \text{Te-234}$
 D) $\text{U-238} \rightarrow \text{Ba-234}$

16. Which statement best shows the relationship in a fission reaction?

A) Energy emitted splits into lighter nuclei.
 B) Light nuclei fission into heavier nuclei.
 C) Energy is released and has visible effects on the nucleus.
 D) Energy is absorbed and causes nuclear fission and fission.

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R Bogdan

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