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Client / Matter

Able, Barry (0002-001)

Matter Details

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Client name: **Able, Barry B.**

Matter number: 0002-001

Description: Able v. State of Florida

Opened: 08/19/12

Type of Law: Federal or State Tax (Personal)

Lawyer name: Robert S. Brown

Rate: \$

Notes:

Related Contacts

S. Adams, Ted 014-000-0012

S. Knight, Tom 014-000-1234

S. Anderson,

0000000001

Events

Add Items

Set Location (Auto-Set)

- Tuesday, 08/20/12
- 10:00 am
- Client Meeting in Office
- File Motion (Extension of Time)
- Get expert affidavit if needed
- Check to see if answer filed or if in default
- File Notice of Hearing
- Discovery due
- Send Interrogatories and other discovery
- Update client on meeting
- Draft Notice of Hearing
- Please get the docs ready for the production
- There is a lot of favorable precedent, but the highly contested

Recent Activity

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Phone Call 01/10/12 02/10/12

Phone Call 01/20/12 10/10/12

Phone Call 01/10/12 04/20/12

Phone Call 01/10/12 02/20/12

Time Entry 08/10/12

Account Details

Client Ledger

Receivable \$7548.79

Unbilled Time \$2325.00

Unbilled Disbursement \$0.00

Trust Balance \$0401.25

Last Time Entry 08/20/12

Last Bill 08/20/12

Last Payment 10/02/12

Last Post Due Notice 02/02/12

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web give results of all calculations in blue cells you must put your answers in the indicated cells as the grade book pulls your data from those cells do not put any units in the number cells use the unit boxes given graph tab transfer from data page the temp and concentration of salt in units of g salt 100 g water

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web use the solubility curve to answer these questions determine how much potassium nitrate will dissolve in 20 g of water at 40 c 200 cm³ of saturated lead ii nitrate solution was prepared at a temperature of 90 c

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web analysis construct a solubility curve using the mass of solute that will dissolve in 100 g of water at a specified temperature in this lab 5 ml of water was used assume 1 0 g ml of water take the collected data make it proportional to get the solubility of the kno3 graph temperature vs solubility using a best fit curve

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