

## Section Review 8.1

### Part A Completion

1. compound
2. molecular
3. nonmetals
4. diatomic
5. molecular formula
6. low
7. high
8. atoms
9. structure

### Part B True-False

10. ST
11. NT
12. AT
13. NT
14. ST

### Part C Matching

15. e
16. a
17. d
18. b
19. c

### Part D Questions and Problems

20. molecular compound
21. a. 4 carbon atoms, 10 hydrogen atoms  
b. 6 carbon atoms, 5 hydrogen atoms, 1 fluorine atom
22. a. molecule  
b. atom  
c. molecule  
d. atom  
e. molecule

## Section Review 8.2

### Part A Completion

1. stable electron
2. covalent
3. shared
4. single
5. unshared pairs
6. double/triple
7. coordinate covalent bond
8. Energy
9. bond dissociation energy
10. resonance structure

### Part B True-False

11. NT
12. NT
13. AT
14. AT
15. ST
16. NT

### Part C Matching

17. e
18. a
19. b
20. c
21. d

## Part D Questions and Problems

22. a.  $\text{Br}:\text{Br}:$
- b.  $\text{H}:\text{C}::\text{N}:$
- c.  $\begin{array}{c} \text{H} \\ \diagup \quad \diagdown \\ \text{H}:\text{N}:\text{H} \\ | \\ \text{H} \end{array}$

## Section 8.3

### Part A Completion

1. molecular orbitals
2. bonding orbital
3. lower
4. sigma or  $\sigma$
5. pi or  $\pi$
6. three-dimensional
7. VSEPR theory
8. orbital hybridization

### Part B True-False

9. AT
10. ST
11. NT
12. AT
13. ST
14. AT

### Part C Matching

15. d
16. e
17. a
18. b
19. c

### Part D Questions and Problems

20. Reading from left to right:  
 $sp^3$   $sp^2$   $sp^2$   $sp$   $sp$   $sp^3$

## Section 8.4

### Part A Completion

1. equally
2. nonpolar
3. unequally
4. polar
5. electronegativities
6. dipole interactions
7. hydrogen bond
8. electronegative
9. oxygen, nitrogen, or fluorine

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**Emilie Sanchez**



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